

## DIMERIZATION OF (E)-2-BENZENESULFONYL-1,3-ALKADIENES. THE BENZENESULFONYL GROUP EXERTS ACYCLIC STEREOCONTROL.

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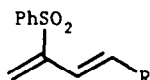
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**Abstract.** Mild dehydration of 2-benzenesulfonyl allylic alcohols **8** afforded a series of (E)-2-benzenesulfonyl-1,3-alkadienes **1** which dimerized to functionalized vinylcyclohexenes **3** in regio- and stereoredefined manner. The configuration at C-3 in **3e** was determined by X-ray crystallography.

**Introduction.** 2-Benzenesulfonyl-1,3-alkadienes **1** have hardly been investigated. Recently, Chou and his coworkers<sup>1</sup> have obtained parent **1a** (R = H)

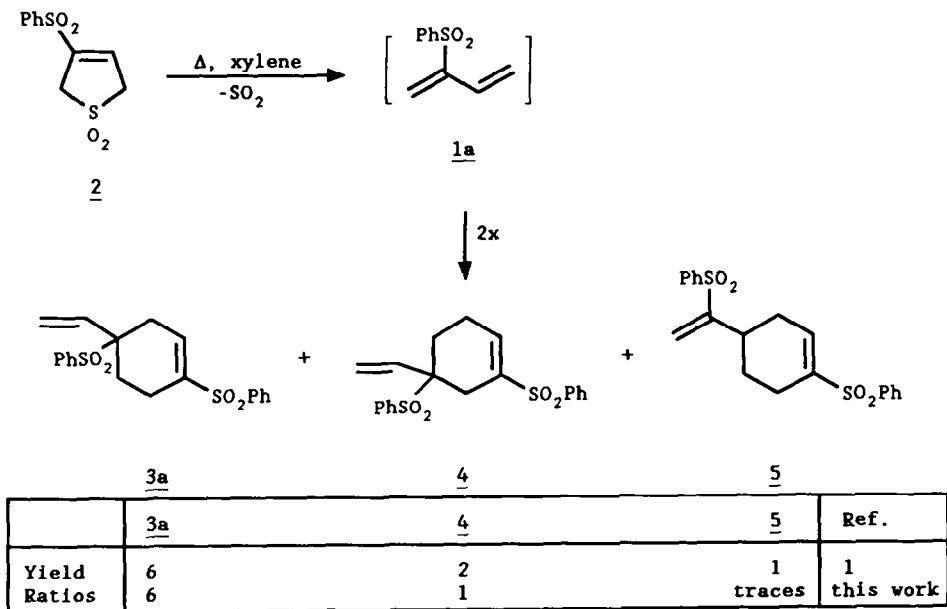


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from 3-sulfolene **2** and shown that it dimerizes *in situ*, affording three isomers, assigned structure **3a** (*para* isomer), **4** (*meta* isomer) and **5** (Scheme 1). Bäckvall et al. have prepared inter al. **1b**<sup>2a</sup> (R = Me) via sulfonylmercuration of 1,3-pentadiene and described one dimerization product (*para*, stereochemistry unknown).<sup>2b</sup> We here report a new approach to 2-benzenesulfonyl-1,3-alkadienes (**1**). The compounds are of interest in

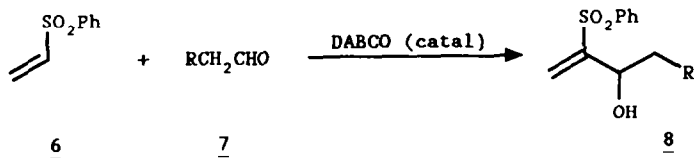
stereoselective cycloadditions and because of the synthetic versatility of the benzenesulfonyl group.<sup>3</sup>

**Scheme 1.**

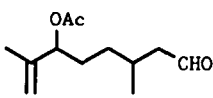


**Results.** Phenyl vinyl sulfone (**6**) and aldehydes **7** were coupled to 2-benzenesulfonyl substituted allylic alcohols **8** in the presence of 1,4-diazabicyclo[2.2.2]octane<sup>4-6</sup> (DABCO) (Scheme 2 and Table 1).

**Scheme 2. Preparation of Benzenesulfonyl Substituted Allyl Alcohols **8****

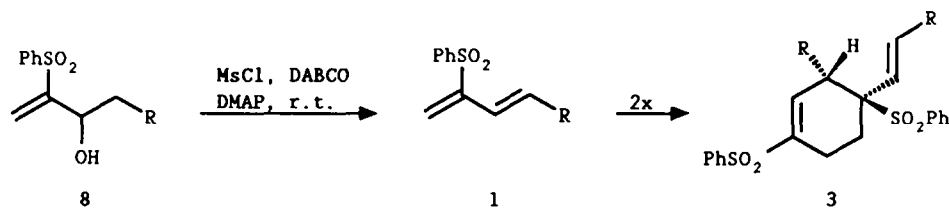


**Table 1.** 2-Benzenesulfonyl-3-hydroxy-1-alkenes (**8a-f**) Prepared

	Aldehyde <sup>a</sup> RCH <sub>2</sub> CHO ( <b>7</b> )	Reaction time	Isolated Yield <sup>b</sup> [%]
<b>a</b>	CH <sub>3</sub> CHO	10d	81
<b>b</b>	CH <sub>3</sub> CH <sub>2</sub> CHO	11d	66 <sup>c</sup>
<b>c</b>	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CHO	4w	60
<b>d</b>	(CH <sub>3</sub> ) <sub>2</sub> CHCH <sub>2</sub> CHO	11w	65
<b>e</b>	Ph(CH <sub>2</sub> ) <sub>2</sub> CHO	3w	46 <sup>d</sup>
<b>f</b>		8w	47

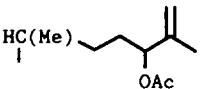
<sup>a</sup>1 - 5 eq of aldehyde **7** was used with respect to phenyl vinyl sulfone **6**. <sup>b</sup>Alcohols **8b-f** were purified by column chromatography, alcohol **8a** was distilled (Kugelrohr). <sup>c</sup>Refers to 50% conversion of **6**. <sup>d</sup>Refers to 78% conversion of **6**.

E-selective dehydration of **8** was accomplished with methanesulfonyl chloride and DABCO<sup>7</sup> at room temperature in the presence of catalytic 4-dimethylaminopyridine (DMAP)<sup>8</sup>, yielding the desired 1,3-alkadienes **1**, which partially dimerized to **3** under the conditions of the reaction and even during workup (Scheme 3). However, chromatographic separation of the respective pairs of **1b-e** and **3b-e** was feasible (silica gel), giving the series of monomers **1b-e** (Table 2), which were characterized in full by <sup>1</sup>H

**Scheme 3.** Dehydration of 2-Benzenesulfonyl Allylic Alcohols **8**

and <sup>13</sup>C NMR (Table 3). Preparation of bulky monomer **1f**, derived from isopulegol (**7f**) succeeded without formation of dimer **3f**. However, complete dimerization ensued when **1f** (neat) was kept for 3 weeks at room temperature.

**Table 2.** Isolation of 2-Benzenesulfonyl-1,3-alkadienes (**1a-f**) and their Dimers (**3a-f**)

	R	<b>1</b> [%]	<b>3</b> [%]	Total Yield [%]
<b>a</b>	H	46 <sup>a</sup>	32 <sup>a</sup>	78
<b>b</b>	CH <sub>3</sub>	39	39	78
<b>c</b>	CH <sub>2</sub> CH <sub>2</sub>	36	45	81
<b>d</b>	CH(CH <sub>3</sub> ) <sub>2</sub>	53	18	71
<b>e</b>	CH <sub>2</sub> Ph	56	14	70
<b>f</b>		86	--	86

<sup>a</sup>Facile dimerization (see text).**Table 3.** NMR Data of 2-Benzenesulfonyl-1,3-alkadienes (**1**) (CDCl<sub>3</sub>, TMS,  $\delta$  ppm)

<b>1</b>	H-1 <sub>cis</sub> - $\Sigma$	H-1 <sub>trans</sub> - $\Sigma$	H-3	H-4(Z)	H-4(E)	H-5	H-6
<b>a</b>	6.37	6.01	6.35	5.68	5.32	-	-
<b>b</b>	6.23	5.87	6.02	6.17	-	1.73	-
<b>c</b>	6.23	5.88	5.99	6.18	-	2.08	0.95
<b>d</b> <sup>a</sup>	6.20	5.88	5.97	6.10	-	2.46	0.92
<b>e</b>	6.27	5.87	6.06	6.26	-	3.37	-
<b>f</b>	6.26	5.88	5.98 <sup>b</sup>	-	-	c	c

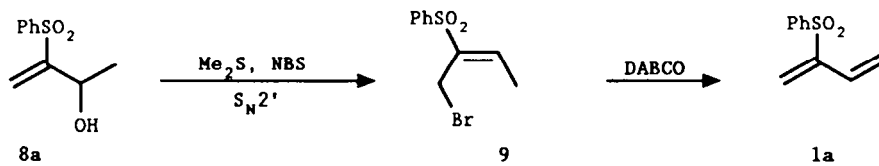
<sup>a</sup>In CD<sub>2</sub>Cl<sub>2</sub>    <sup>b</sup>AB part of ABX    <sup>c</sup>No assignment     $\Sigma$  = PhSO<sub>2</sub>

<b>1</b>	C-1	C-2	C-3	C-4	C-5	C-6
<b>b</b>	121.80	148.20	133.72	121.92	18.64	-
<b>c</b>	121.73	148.36	138.93	119.76	26.01	12.81
<b>d</b> <sup>a</sup>	122.02	148.91	145.85	118.33	31.90	22.97
<b>e</b>	122.36	147.92	137.01	121.95	39.01	-
<b>f</b>	122.22	148.18	143.89	119.72	37.11	30.07

<sup>a</sup>In CD<sub>2</sub>Cl<sub>2</sub>

In accord with the report of Chou<sup>1</sup> parent **1a** could only be handled in dilute solution, dimerization to **3a**, **4** and **5** being fast at room temperature. 2-Benzenesulfonyl-1,3-butadiene (**1a**) was also prepared from **8a** via **9** (Scheme 4).

**Scheme 4. Alternative Route to 1a**

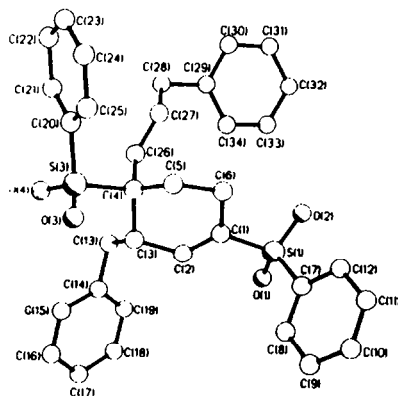


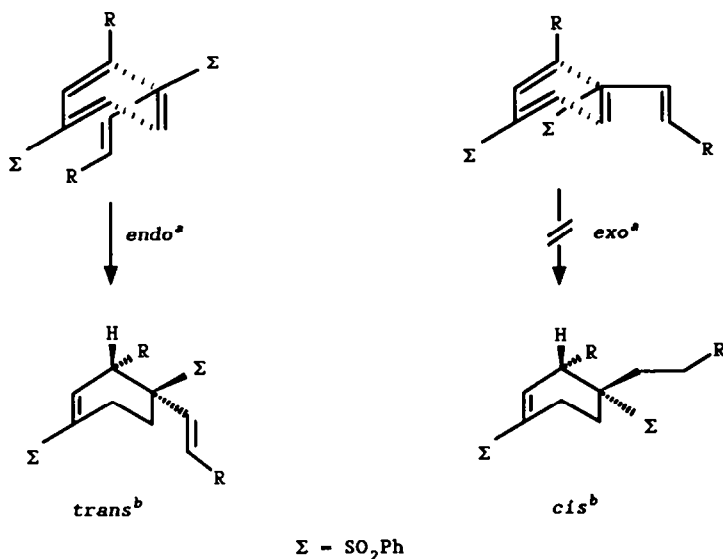
A somewhat higher regioselectivity in favour of *para* oriented dimer **3a** was obtained in our work (Scheme 1) by using the milder method for preparing **1**.

**Regioselectivity and Stereochemistry of Dimerization.** Unlike the dimerization of parent **1a** which was regiorandom (Scheme 1), 4-alkylated 2-benzenesulfonyl-1,3-dienes **1b-e** dimerized to *para* isomers **3b-e** exclusively, which crystallized on standing at  $-20^{\circ}\text{C}$ . Crystallization of dimer **3f** was not successful.

Dimerization was not only regioselective. The  $^{13}\text{C}$  spectra showed just one of two possible diastereomeric dimers. Spectroscopic data did not allow clear assignment of the configuration at C-3. However, the X-ray crystal structure determination of **3e** revealed the bulky benzenesulfonyl group to be *trans* to the vicinal benzyl group (Fig. 1). Therefore, all members of the series **3b-f** have been formed in an *endo* sense, irrespective of the size of the alkyl group R (Scheme 5).

**Figure 1. X-ray Crystal Structure of 3e**

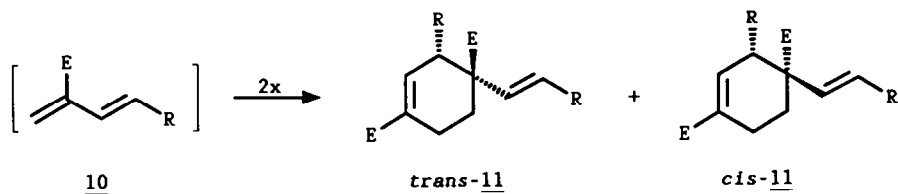


**Scheme 5. Endo versus Exo Dimerization**

<sup>a</sup>*endo* and *exo* refer to position of alkenyl group with respect to roof-like transition state of Diels-Alder dimerization.

<sup>b</sup>*trans* and *cis* refer to the relationship of benzenesulfonyl group at C-2 and alkyl group R at C-3.

Previously, we have shown<sup>7a</sup> that a series of 2-methylene-3-alkenoic esters **10** ( $\text{E} = \text{CO}_2\text{Me}$ ), which were generated at room temperature, dimerized sponta-



aneously to *trans*- and *cis*-**11**. Similarly, the related group of acetyl-activated dienes **10** ( $\text{E} = \text{COMe}$ ) dimerized to a mixture of diastereomers. Both classes of compounds (**10**,  $\text{E} = \text{CO}_2\text{Me}$  and  $\text{COMe}$ ) were *more* reactive than the sulfonyl activated dienes **1**, whereas the steric course of dimerization was *less* selective (Table 4).

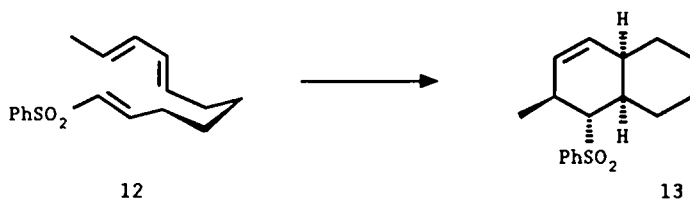
Complete diastereocontrol in the self-dimerization of sulfonyl dienes **1** appears to be the consequence of the bulk of the benzenesulfonyl group (A value 2.50)<sup>9a</sup> and the diminished reactivity of **1** compared with ester **10** ( $\text{E} = \text{CO}_2\text{Me}$ ,  $\text{COMe}$ ).

**Table 4.** Diastereocontrol of Dimerization as a Function of Acceptor Group

Diene	Acceptor Group E	A value of Acceptor	Diastereoselectivity <i>trans</i> : <i>cis</i>
1	PhSO <sub>2</sub>	2.50 <sup>a</sup>	3; 100%
10	CO <sub>2</sub> Me	0.60 <sup>b</sup>	11; ~5:1 to 10:1 <sup>c</sup>
10	COMe	Me(1.70) > MeO(0.60) <sup>b</sup>	11; 2:1 to 4:1 <sup>d</sup>

<sup>a</sup>Ref. 9a. <sup>b</sup>Ref. 9b. <sup>c</sup>Ref. 7a. <sup>d</sup>Ref. 7d.

It is also of interest that the intramolecular Diels-Alder reaction of **12** affords diastereomer **13**, in which the sterically demanding benzenesulfonyl group adopts the equatorial position *trans* to the neighbouring methyl group.<sup>10</sup> Presumably, the alternative *cis* transition state entails unfavourable repulsion of these two groups.



In conclusion, because of its size and electronic properties the benzenesulfonyl group exerts useful regio and stereo control in a pericyclic reaction. Starting from conformationally mobile, acyclic reactants **1** we have obtained functionalized cyclohexenes **3** with 100% regio- and 100% diastereocontrol, even before recrystallization. Diastereocontrol is lost partially if benzenesulfonyl is replaced by an ester grouping (**10**; E = CO<sub>2</sub>Me)<sup>7a</sup> and even more so, by an acetyl group (**10**; E = COMe)<sup>7b</sup>.

#### EXPERIMENTAL

For the preparation and spectral data of  $\alpha$ -hydroxyalkylated vinylsulfones (**8a-e**) see ref. 4c.

**8-Acetoxy-2-benzenesulfonyl-5,9-dimethyl-3-hydroxy-1,9-decadiene (8f).** Aldehyde **7f**<sup>11</sup> (3.3 g, 15.5 mol) and sulfone (**6**) (2.6 g, 15.5 mmol) were mixed and 1,4-diazabicyclo[2.2.2]octane (170 mg, 1.55 mmol) was added. After 8 w at room temperature, the crude product was purified by flash chromatography on silica (ether/ light petroleum, 2 : 1) to give **8f** (2.8 g,

47%, diastereomeric mixture), as a viscous oil. IR (CHCl<sub>3</sub>)  $\nu$  3540, 3020, 2960, 1730, 1450, 1370, 1305, 1250, 1140, 1080, 1025, 965, 910, 710, 690, 590 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  7.98 - 7.51 (m, 5 H, H<sub>arom</sub>), 6.41, 6.39 (d, <sup>2</sup>J = 0.75 Hz, 1 H, PhSO<sub>2</sub>C=CHH<sub>cis</sub>), 6.14/6.12 (bs, 1 H, PhSO<sub>2</sub>C=CHH<sub>trans</sub>), 5.09/5.07 (t, <sup>3</sup>J = 7 Hz, 1 H, CHOAc), 4.89 (m, 2 H, CH<sub>3</sub>C=CH<sub>2</sub>), 4.42 (m, 1 H, CHOH), 3.21 (bs, 1 H, OH), 2.06/2.05 (s, 3 H, COCH<sub>3</sub>), 1.70, 1.68 (s, 3 H, CCH<sub>3</sub>), 1.64 - 0.87 (m, 7 H), 0.82/0.76 (d, <sup>3</sup>J = 6 Hz, 3 H, CHCH<sub>3</sub>); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>)  $\delta$  170.39/170.32 (s, C=O), 154.40/153.9 (s, PhSO<sub>2</sub>C=), 143.04 (s, CH<sub>3</sub>C=CH<sub>2</sub>), 139.45/139.39 (s, C<sub>arom</sub>), 133.73, 129.33, 129.23 (d, C<sub>arom</sub>), 124.58/124.37 (t, PhSO<sub>2</sub>C=CH<sub>2</sub>), 113.01/112.9 (t, CH<sub>3</sub>C=CH<sub>2</sub>), 66.99/66.57 (d, CHOH), 43.71/43.45 (d, CHOAc), 32.94, 31.56, 29.16 (t, 3 CH<sub>2</sub>), 29.86 (d, CHCH<sub>3</sub>), 21.17 (q, COCH<sub>3</sub>), 19.89/19.7 (q, CCH<sub>3</sub>), 18.00/17.96 (q, CHCH<sub>3</sub>); mass spectrum (70 eV, 130°C), *m*/*i* (relative intensity) 380 (0, M<sup>+</sup>), 348 (1), 333 (2), 319 (5), 196 (25), 17 (39), 178 (33), 161 (34), 125 (62), 109 (50), 108 (50), 93 (57), 81 (49), 77 (71), 44 (100). Exact mass calcd for C<sub>12</sub>H<sub>20</sub>O<sub>2</sub> 196.1464, Found 196.1462

**General Procedure for the Dehydration of 2-Benzenesulfonyl-3-hydroxy-1-alkenes (8a-f).** 1,4-Diazabicyclo[2.2.2]octane (3 eq) and a catalytic amount of *p*-(dimethylamino)pyridine were added to a solution of allylic alcohol (1 eq) in dry CH<sub>2</sub>Cl<sub>2</sub> (1.5 mL/mmol **8**) under nitrogen. The mixture was cooled to 0°C and methanesulfonyl chloride (1.2 eq in 1.5 mL CH<sub>2</sub>Cl<sub>2</sub>/mmol **8**) was dropped in slowly by syringe. The formation of a colorless precipitate was observed after a short time. The mixture was stirred at room temperature for the indicated reaction time, diluted with CH<sub>2</sub>Cl<sub>2</sub>, washed with water (3x), and dried (MgSO<sub>4</sub>). After filtration and evaporation of the solvent the residue was flash chromatographed on silica gel as detailed.

**Dehydration of 8a.** Reaction of alcohol **8a** (3.18 g, 15 mmol) with DABC (5.04 g, 45 mmol), cat. DMAP and methanesulfonyl chloride (2.06 g, 1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (45 mL) for 6 h (TLC) gave 2.37 g of crude brown oil. Separation of the component with the higher R<sub>F</sub>-value by chromatography (ether/light petroleum, 2 : 1) yielded 1.33 g (46%) of a yellowish oil which consisted of compounds **1a**, **3a**, **4** in a 5 : 10 : 3 ratio by the time the <sup>1</sup>H NMR spectrum was recorded. The second fraction crystallized to give 0.93 g (32%) of *para*-dimer **3a**, slightly contaminated with dimer **5**. For spectroscopic data of **1a** see ref. 2b, for **3a**, **4**, **5** see ref. 1a. Alternatively, formation of diene **1a** was followed by <sup>1</sup>H NMR by reaction of bromide **9**<sup>5</sup> (30 mg, 0.11 mmol in 1 mL of CDCl<sub>3</sub>) with two equivalents of DABC at room temperature. In solution (0.11 M), diene **1a** was stable for at least 8 h.

**2-Benzenesulfonyl-1,3-pentadiene (1b) and 1,4-Bis(benzenesulfonyl)-3-methyl-4-(1-propenyl)-1-cyclohexene (3f).** Alcohol **8b** (230 mg, 1 mmol) was



dehydrated within 18 h by action of DABCO (340 mg, 3 mmol), cat. DMAP and methanesulfonyl chloride (140 mg, 1.2 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (3 mL). After aqueous workup, chromatography (ether/light petroleum, 2 : 1) yielded 80 mg (39%) of monomeric diene **1b** as a colorless liquid (first fraction). For  $^1\text{H}$  NMR spectral data see ref. 2b.  $^{13}\text{C}$  NMR (50 MHz, APT, subtractive)  $\delta$  148.20 (s,  $\text{C}=\text{CH}_2$ ), 139.42 (s,  $\text{C}_{\text{arom}}$ ), 133.72 (d,  $=\text{CCH}$ ), 133.40, 129.10, 127.94 (d,  $\text{C}_{\text{arom}}$ ), 121.92 (d,  $\text{CCH}=\text{CH}$ ), 121.80 (t,  $\text{C}=\text{CH}_2$ ), 18.64 (q,  $\text{CH}_3$ ). The second fraction crystallized from the above solvent mixture to give dimer **3b** (50 mg, 24%), colorless needles, mp 145 - 147°C (ether/light petroleum). For  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectral data see ref. 2b. IR (KBr)  $\nu$  3080, 2960, 1650, 1580, 1450, 1300, 1150, 1090, 1070, 760, 720, 690  $\text{cm}^{-1}$ ; mass spectrum (70 eV, 160°C),  $m/z$  (relative intensity) 416 (0,  $\text{M}^+$ ), 275 (36), 259 (6), 150 (11), 149 (11), 133 (41), 125 (100), 105 (20), 91 (20), 77 (36). Exact mass calcd for  $\text{C}_{16}\text{H}_{19}\text{O}_2\text{S}$  275.1105, found 275.1104. Anal. Calcd for  $\text{C}_{22}\text{H}_{24}\text{O}_4\text{S}_2$ : C, 63.44; H, 5.81. Found: C, 62.91; H, 5.80.

**2-Benzenesulfonyl-1,3-hexadiene (1c) and 1,4-Bis(benzenesulfonyl)-4-(1-butenyl)-3-ethyl-1-cyclohexene (3c)**. Conversion of **8c** (240 mg, 1 mmol) with the same amounts of reactants as described for the preparation of **1b** and **3b** afforded, within 18 h, 80 mg (36%) of diene **1c**, colorless liquid after chromatography (ether/light petroleum, 2 : 1).  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ , subtractive)  $\delta$  7.92 - 7.47 (m, 5 H,  $\text{H}_{\text{arom}}$ ), 6.23 (s, 1 H,  $=\text{CHH}_{\text{cis}}$ ), 6.18 (dt,  $^3J = 16 \text{ Hz}/^3J = 6 \text{ Hz}$ , 1 H,  $\text{CCH}=\text{CH}$ ), 5.99 (dq,  $^3J = 16 \text{ Hz}/^4J = 1 \text{ Hz}$ , 1 H,  $=\text{CCH}$ ), 5.88 (s, 1 H,  $\text{CHH}_{\text{trans}}$ ), 2.08 (dq,  $^3J = 6 \text{ Hz}/^3J = 7 \text{ Hz}$ , 2 H,  $\text{CH}_2\text{CH}_3$ ), 0.95 (t,  $^3J = 7 \text{ Hz}$ , 3 H,  $\text{CH}_3$ );  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  148.36 (s,  $\text{C}=\text{CH}_2$ ), 138.93 (d,  $=\text{CCH}$ ), 138.87 (s,  $\text{C}_{\text{arom}}$ ), 133.86, 129.04, 128.00 (d,  $\text{C}_{\text{arom}}$ ), 121.73 (t,  $\text{C}=\text{CH}_2$ ), 119.76 (d,  $=\text{CCHCH}$ ), 26.01 (t,  $\text{CH}_2\text{CH}_3$ ), 12.81 (q,  $\text{CH}_3$ ). Further elution gave dimer **3c** [(60 mg, 45%, colorless needles, mp 139 - 140°C (ether/light petroleum)] as a second fraction. IR ( $\text{CHCl}_3$ )  $\nu$  3040, 2980, 1650, 1450, 1300, 1150, 1080, 690  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.87 - 7.43 (m, 10 H,  $\text{H}_{\text{arom}}$ ), 7.01 (m, 1 H,  $\text{PhSO}_2\text{C}=\text{CH}$ ), 5.51 (d,  $^3J = 16 \text{ Hz}$ , 1 H,  $\text{CCH}=\text{CH}$ ), 5.33 (dt,  $^3J = 16 \text{ Hz}/^3J = 6 \text{ Hz}$ , 1 H,  $\text{CCH}=\text{CH}$ ), 2.98 (m, 1 H,  $\text{PhSO}_2\text{C}=\text{CHCH}$ ), 2.47 - 1.98 (m, 4 H,  $(\text{CH}_2)_2$ ), 1.96 (m, 4 H,  $\text{CH}_2\text{CH}_3$ ), 1.11 (t,  $^3J = 7.5 \text{ Hz}$ , 3 H,  $\text{CHCH}_2\text{CH}_3$ ), 0.83 (t,  $^3J = 7.5 \text{ Hz}$ , 3 H,  $=\text{CHCH}_2\text{CH}_3$ );  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  139.67, 139.04 (d,  $\text{PhSO}_2\text{C}=\text{CH}$ ,  $\text{CCH}=\text{CH}$ ), 139.12, 138.03 (s,  $\text{C}_{\text{arom}}$ ), 135.40 (s,  $\text{PhSO}_2\text{C}=\text{}$ ), 133.83, 133.32, 130.73, 129.11, 128.44, 127.98 (d,  $\text{C}_{\text{arom}}$ ), 120.68 (d,  $\text{CCH}=\text{CH}$ ), 68.46 (s,  $\text{CCH}=\text{CH}$ ), 40.86 (d, C-3), 26.14, 25.82 (t,  $(\text{CH}_2)_2$ ), 23.53 (t,  $\text{CH}=\text{CHCH}_2$ ), 20.62 (t,  $\text{CHCH}_2\text{CH}_3$ ), 12.93, 12.03 (q,  $\text{CH}_3$ ); mass spectrum (70 eV, 50°C),  $m/z$  (relative intensity) 444 (1,  $\text{M}^+$ ), 227 (15), 196 (40), 125 (32), 115 (82), 77 (100), 71 (73). Anal. Calcd for  $\text{C}_{24}\text{H}_{28}\text{O}_4\text{S}_2$ : C, 64.84; H, 6.35. Found: C, 64.03; H, 6.33.

**2-Benzenesulfonyl-5-methyl-1,3-hexadiene (1d) and 1,4-Bis(benzenesulfonyl-3-isopropyl-4-[(3-methyl)-1-butenyl]-1-cyclohexene (3d).** Reaction of alcohol **8d** (310 mg, 1.2 mmol) with DABCO (400 mg, 3.6 mmol), cat. DMAP and methanesulfonyl chloride (165 mg, 1.44 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (4 mL) for 24 h yielded two fractions on chromatography (ether/light petroleum, 1 : 2). The first fraction was liquid diene **1d** (150 mg, 53%).  $^1\text{H}$  NMR (200 MHz  $\text{CD}_2\text{Cl}_2$ , subtractive)  $\delta$  7.92 - 7.42 (m, 5 H,  $\text{H}_{\text{arom}}$ ), 6.20 (s, 1 H, = $\text{CHH}_{\text{cis}}$ ), 6.11 (dd,  $^3J = 16\text{ Hz}/^3J = 6\text{ Hz}$ , 1 H, = $\text{CCHCH}$ ), 5.97 (dt,  $^3J = 1\text{ Hz}/^4J = 1\text{ Hz}$ , 1 H, = $\text{CCHCH}$ ), 5.88 (s, 1 H,  $\text{CHH}_{\text{trans}}$ ), 2.46 (m, 1 H,  $\text{CH}(\text{CH}_3)_2$ ), 0.92 (d,  $^3J = 7\text{ Hz}$ , 6 H,  $\text{CH}_3$ );  $^{13}\text{C}$  NMR (50 MHz,  $\text{CD}_2\text{Cl}_2$ , subtractive)  $\delta$  148.91 (s,  $\text{C}=\text{CH}_2$ ), 145.84 (d,  $\text{CCH}=\text{CH}$ ), 139.95 (s,  $\text{C}_{\text{arom}}$ ), 133.70, 129.37, 128.31 (d,  $\text{C}_{\text{arom}}$ ), 122.02 (t,  $\text{C}=\text{CH}_2$ ), 118.33 (d,  $\text{CCH}=\text{CH}$ ), 31.9 (d,  $\text{CH}(\text{CH}_3)_2$ ), 21.71 (q,  $\text{CH}_3$ ). Further elution gave 50 mg (18%) of diene **3d** as a colorless oil, which crystallized on standing at  $-20^\circ\text{C}$  (needles mp  $135 - 137^\circ\text{C}$ , ether). IR ( $\text{CHCl}_3$ )  $\nu$  3020, 2970, 1445, 1320, 1300, 1190, 1150, 910, 690  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.93 - 7.40 (m, 10 H,  $\text{H}_{\text{arom}}$ ), 7.09 (m, 1 H,  $\text{PhSO}_2\text{C}=\text{CH}$ ), 5.71 (dd,  $^3J = 17\text{ Hz}/^4J = 1.5\text{ Hz}$ , 1 H,  $\text{CCHCH}$ ), 5.20 (dd,  $^3J = 17\text{ Hz}/^3J = 7\text{ Hz}$ , 1 H,  $\text{CCHCH}$ ), 3.37 (m, 1 H,  $\text{PhSO}_2\text{C}=\text{CHCH}$ ), 2.53 - 2.12 (m, 4 H,  $(\text{CH}_2)_2$ ), 1.85 (m, 2 H,  $\text{CH}(\text{CH}_3)_2$ ), 1.21 (d,  $^3J = 7\text{ Hz}$ ), 0.86 (d,  $^3J = 6.5\text{ Hz}$ ), 0.84 (d,  $^3J = 6.5\text{ Hz}$ ), 0.72 (d,  $^3J = 7\text{ Hz}$ ) (each 3 H,  $\text{CH}_3$ );  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  142.54 (d,  $\text{PhSO}_2\text{C}=\text{CH}$ ), 139.30 (s,  $\text{C}_{\text{arom}}$ ), 136.35 (d,  $\text{CCH}=\text{CH}$ ), 134.72 (s,  $\text{PhSO}_2\text{C}=\text{CH}$ ), 133.78, 133.25, 130.65, 129.08, 128.31, 127.93 (d,  $\text{C}_{\text{arom}}$ ), 121.89 (d,  $\text{CCH}=\text{CH}$ ), 68.16 (s,  $\text{CCH}=\text{CH}$ ), 42.58 (d,  $\text{PhSO}_2\text{C}=\text{CHCH}$ ), 31.70, 28.62 (d,  $\text{CH}(\text{CH}_3)_2$ ), 25.39, 20.30 (t,  $(\text{CH}_2)_2$ ), 22.97, 21.91, 21.52, 18.25 (q,  $\text{CH}_3$ ); mass spectrum (70 eV,  $180^\circ\text{C}$ ),  $m/z$  (relative intensity) 472 (0,  $\text{M}^+$ ), 331 (23), 288 (36), 287 (35), 275 (10), 263 (16), 189 (100), 125 (89), 91 (40), 78 (45). Exact mass calcd for  $\text{C}_{18}\text{H}_{27}\text{O}_2\text{S}$  331.1732, found 331.1731. Anal. Calcd for  $\text{C}_{26}\text{H}_{32}\text{O}_4\text{S}_2$ : C, 66.07; H, 6.82. Found: C, 65.95; H, 6.83.

**2-Benzenesulfonyl-5-phenyl-1,3-pentadiene (1e) and 3-Benzyl-1,4-bis(benzenesulfonyl)-4-(3-phenyl-1-propenyl)-1-cyclohexene (3e).** Alcohol **8e** (140 mg, 0.5 mmol) was allowed to react with DABCO (170 mg, 1.5 mmol), cat. DMAP and methanesulfonyl chloride (70 mg, 0.6 mmol) in dry  $\text{CH}_2\text{Cl}_2$  (3 mL) for 21 h, following the general procedure. Chromatography (ether/light petroleum, 2 : 1) gave 80 mg (56%) of liquid diene **1e** as first eluted compound. IR ( $\text{CHCl}_3$ )  $\nu$  3010, 2920, 1445, 1305, 1145, 910, 690  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ , subtractive)  $\delta$  7.88 - 7.40 (m, 5  $\text{H}_{\text{arom}}$ ), 7.40 - 6.91 (m, 5 H,  $\text{H}_{\text{arom}}$ ), 6.27 (s, 1 H, = $\text{CHH}_{\text{cis}}$ ), 6.26 (dt,  $^3J = 15.5\text{ Hz}/^3J = 6.5\text{ Hz}$ , 1 H,  $\text{CCH}=\text{CH}$ ), 6.06 (dd,  $^3J = 15.5\text{ Hz}/^4J = 1\text{ Hz}$ , 1 H,  $\text{CCH}=\text{CH}$ ), 5.87 (s, 1 H,  $\text{CHH}_{\text{trans}}$ ), 3.37 (d,  $^3J = 6.5\text{ Hz}$ , 2 H,  $\text{CH}_2$ );  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ , AP subtractive)  $\delta$  147.92 (s,  $\text{C}=\text{CH}_2$ ), 139.15 (s,  $\text{C}_{\text{arom}}$ ), 137.01 (d, = $\text{CCH}$

134.35 (s, C<sub>arom</sub>), 133.27, 128.98, 128.38, 127.92, 126.29 (d, C<sub>arom</sub>), 122.36 (t, C=CH<sub>2</sub>), 121.95 (d, =CCH=CH), 39.01 (t, CH<sub>2</sub>); mass spectrum (70 eV, r.t.), *m/z* (relative intensity) 284 (0, M<sup>+</sup>), 283 (6), 141 (100), 127 (66), 77 (30). The second fraction was oily dimer **3e** (20 mg, 14%), which crystallized from ether on prolonged standing at -20°C (colorless needles, mp 155 - 156°C). IR (CHCl<sub>3</sub>)  $\nu$  3020, 1145, 1305, 1145, 690 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  7.81 - 7.02 (m, 15 H, H<sub>arom</sub>), 6.78 (m, 1 H, PhSO<sub>2</sub>C=CH), 5.79 (d, <sup>3</sup>*J* = 16 Hz, 1 H, CCH=CH), 5.46 (dt, <sup>3</sup>*J* = 16 Hz/<sup>3</sup>*J* = 7 Hz, 1 H, CCH=CH), 3.83 ((dd, *J* = 6.5/3 Hz), 3.56 (m, 2 H, diastereotopic, -CHCH<sub>2</sub>Ph), 3.37 (d, <sup>3</sup>*J* = 7 Hz, 2 H, =CHCHPh), 2.28 - 1.80 (m, 5 H, PhSO<sub>2</sub>C=CHCH, (CH<sub>2</sub>)<sub>2</sub>); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>)  $\delta$  138.69, 138.57, 138.26, 138.10 (s, C<sub>arom</sub>), 138.15, 137.08 (d, PhSO<sub>2</sub>C=CH, CCH=CH), 134.55 (s, PhSO<sub>2</sub>C=CH), 133.85, 133.38, 130.73, 129.42, 129.09, 128.86, 128.73, 128.58, 128.40, 128.05, 126.66, 126.55 (d, C<sub>arom</sub>), 122.66 (d, CCH=CH), 68.11 (s, CCH=CH), 40.97 (d, PhSO<sub>2</sub>C=CHCH), 39.64, 36.68 (t, CH<sub>2</sub>Ph), 25.90, 20.55 (t, (CH<sub>2</sub>)<sub>2</sub>); mass spectrum (70 eV, 210°C), *m/z* (relative intensity) 568 (0, M<sup>+</sup>), 427 (14), 335 (47), 285 (11), 209 (14), 194 (32), 171 (28), 125 (29), 117 (17), 91 (100), 77 (29). Exact mass calcd for C<sub>28</sub>H<sub>27</sub>O<sub>2</sub>S 427.1732, found 427.1732. Anal. Calcd for C<sub>34</sub>H<sub>32</sub>O<sub>4</sub>S<sub>2</sub>: C, 72.80; H, 5.67. Found: C, 71.31; H, 5.75.

**8-Acetoxy-2-benzenesulfonyl-5,9-dimethyl-1,3,9-decatriene (1f) and 3-(4-Acetoxy-1,5-dimethyl-5-hexenyl)-4-(6-acetoxy-3,7-dimethyl-1,7-octadienyl)-1,4-bis(benzenesulfonyl)-1-cyclohexene (3f)**. Alcohol **8f** (220 mg, 0.58 mmol) was dehydrated within 24 h by reaction of DABCO (195 mg, 1.74 mmol), cat. DMAP and methanesulfonyl chloride (80 mg, 0.7 mmol). TLC showed triene **1f** as only product, which was purified by chromatography (ether/light petroleum, 2 : 1) to yield 180 mg (86%) of a colorless oil. IR (CHCl<sub>3</sub>)  $\nu$  2960, 2880, 1730, 1445, 1370, 1320, 1305, 1250, 1145, 1080 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  7.86 - 7.55 (m, 5 H, H<sub>arom</sub>), 6.26 (s, 1 H, PhSO<sub>2</sub>C=CHH<sub>cis</sub>), 5.98 (s, AB-part of ABX-system, 2 H, =CCH=CH), 5.88 (s, 1 H, PhSO<sub>2</sub>C=CHH<sub>trans</sub>), 5.05 (t, <sup>3</sup>*J* = 6 Hz, 1 H, C(H)OAc), 4.87 (m, 2 H, CH<sub>3</sub>C=CH<sub>2</sub>), 2.22 - 2.06 (m, X-part of ABX-system, 1 H, CHCH<sub>3</sub>), 2.06 (s, 3 H, C(=O)CH<sub>3</sub>), 1.64 (s, 3 H, C(=CH<sub>2</sub>)CH<sub>3</sub>), 1.49 - 1.08 (m, 4 H, (CH<sub>2</sub>)<sub>2</sub>), 0.96 (d, <sup>3</sup>*J*<sub>5,11</sub> = 7 Hz, 3 H, CHCH<sub>3</sub>); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>, APT)  $\delta$  170.21 (s, C=O), 148.16 (s, PhSO<sub>2</sub>C), 143.89 (d, =CCH=CH), 142.88 (s, CH<sub>3</sub>C=CH<sub>2</sub>), 139.40 (s, C<sub>arom</sub>), 133.39, 129.04, 127.90 (d, C<sub>arom</sub>), 122.22 (t, PhSO<sub>2</sub>C=CH<sub>2</sub>), 119.72 (d, =CCH=CH), 112.72 (t, CH<sub>3</sub>C=CH<sub>2</sub>), 76.93 (d, OCH), 37.11 (d, CHCH<sub>3</sub>), 31.83 (t, OCHCH<sub>2</sub>), 30.07 (t, OCHCH<sub>2</sub>CH<sub>2</sub>), 21.18 (q, C(=O)CH<sub>3</sub>), 20.08 (q, C(=CH<sub>2</sub>)CH<sub>3</sub>), 18.12 (q, CHCH<sub>3</sub>); mass spectrum (70 eV, r.t.), *m/z* (relative intensity) 362 (2, M<sup>+</sup>), 342 (1), 320 (7), 302 (12), 233 (18), 221 (24), 178 (17), 161 (75), 94 (100), 78 (58), 56 (46), 43 (92). The

dimer **3f** (colorless resin, diastereomeric mixture) was obtained in quantitative yield after keeping neat monomeric triene **1f** at room temperature for about 3 w. IR (CHCl<sub>3</sub>)  $\nu$  3020, 2960, 1730, 1445, 1375, 1320, 1312, 1250, 1150, 1085, 1025 cm<sup>-1</sup>; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.93–7.40 (m, 10 H<sub>arom</sub>), 7.00 (m, 1 H, PhSO<sub>2</sub>C=CH), 5.76–5.58 (m, 2 H, C=CH=CH), 5.2–4.83 (m, 6 H, OCH, CH<sub>3</sub>C=CH<sub>2</sub>), 3.45–3.29 (m, 1 H, PhSO<sub>2</sub>C=CHCH), 2.1–0.67 (m, other H); mass spectrum (70 eV, 220°C), *m/z* (relative intensity) 724 (0, M<sup>+</sup>), 583 (2), 523 (2), 480 (2), 462 (6), 381 (21), 353 (22), (57), 142 (52), 125 (51), 110 (100), 77 (52). Exact mass calcd C<sub>34</sub>H<sub>47</sub>O<sub>6</sub>S 583.3093, found 583.3092.

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**X-Ray Crystallographic Data.** Crystal data and lists of X-ray data for sulfone **3e** (Tables 1–6) have been deposited at the Cambridge Crystallographic Data Centre, U.K.